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Preliminary Communication

A *aihapto-*IMINOACYL DERIVATIVE OF RUTHENIUM(II). SYNTHESIS AND STRUCTURE OF  $Ru[n^2-C(NR)R]C1(CO)(PPh_3)_2$ 

W.R. ROPER,\* G.E. TAYLOR, J.M. WATERS and L.J. WRIGHT Department of Chemistry, University of Auckland, Auckland (New Zealand) (Received July 11th, 1978)

## SUMMARY

The ruthenium iminoacyl complex,  $\operatorname{Ru}[C(NR)R]Cl(CO(PPh_3)_2, (R = p-tolyl),$ which results from a migratory-insertion rearrangement of  $\operatorname{RuRCl(CO)(CNR)(PPh_3)_2}$ is shown by X-ray crystal structure analysis to contain the iminoacyl ligand attached to ruthenium through C and N in an almost symmetrical manner.

We recently reported the five co-ordinate ruthenium(II) aryl, RuRX(CO) (PPh<sub>3</sub>)<sub>2</sub>, I, and the further reaction with CO to form  $RuRX(CO)_2(PPh_3)_2$ , II, (ref. 1). A solution of II was shown to be in equilibrium with the acyl-derivative  $Ru[C(O)R]X(CO)(PPh_3)_2$  for which a *dihapto*-arrangement of the acyl group was suggested. In an extension of this work we describe here reaction of I with CNR and the ensuing migratory-insertion reaction to yield  $Ru[C(NR)R]X(CO)(PPh_3)_2$ .

I, (X = Cl), reacts rapidly with CNR to form the colourless octahedral complex RuRCl(CO)(CNR)(PPh<sub>3</sub>)<sub>2</sub> for which the most likely geometry, based on the known geometry of I, is III (see Scheme). When a solution of III in dichloromethane is heated under reflux the solution becomes orange and the orange crystalline product, V, has no isocyanide  $v_{\rm CN}$  band in the I.R. spectrum but instead has a band at 1654 cm<sup>-1</sup> appropriate for an iminoacyl ligand (see Table for other I.R. data).

A dichloromethane solution of V is in equilibrium with a very small amount of a compound having  $v_{CO}$  and  $v_{CN}$  close to but not identical with those of compound III. This must be one of the two possible isomers of III with *cis* R and CNR ligands, IV.

Compound V does not have the intense colour of other five co-ordinate Ru(II) complexes nor does it readily react with further CO or CNR. It therefore seemed likely that the iminoacyl ligand was occupying two co-ordination sites as has been demonstrated for the iminoacyl ligand in the molybdenum complex,  $Mo[n^2-C(NC_6H_5)CH_3](n^5-C_5H_5)(CO)_2$  (ref. 2). Confirmation was sought through an X-ray structure determination.

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SCHEME: (L =  $PPh_3$ ; R = p-tolyl)

TABLE :	I.R.	data	for	ruthenium(II)	complexes

Compound <sup>a</sup>	$v_{\rm CO}^{\rm b}({\rm cm}^{-1})$	$v_{CN}^{b}(cm^{-1})$	
RuRCl(CO)(CNR)(PPh3)2III	1948	2125	
RURC1 (CO) (CNR) (PPh3) 2IV	1970	2145	
$\operatorname{Ru}[\eta^2 - C(NR)R]C1(CO)(PPh_3)_V$	1902	1654	
$\operatorname{RuCl}_{2}[C(\mathrm{NHR})\mathrm{R}](\mathrm{CO})(\operatorname{PPh}_{3})_{2}^{VI}$	1975	1510	

<sup>a</sup> All compounds have satisfactory elemental analyses and <sup>1</sup>H N.M.R. data consistent with formulations given.

<sup>b</sup> Nujol mulls

 $\operatorname{Ru}[n^2-C(NR)R]Cl(CO)(PPh_3)_2$  crystallizes in the monoclinic system with a = 12.769(8), b = 23.938(2), c = 14.825(2) Å,  $\beta = 104.182(6)^\circ$ , Z = 4, space group  $\operatorname{P2}_1/c$ . Intensity data were collected on a Hilger-Watts four-circle diffractometer using Cu-K<sub>a</sub> radiation and a total of 3688 unique reflections with  $I > 3\sigma(I)$  were recorded. The structure was solved by Patterson and Fourier methods and has been refined to R = 0.052 based on a disordered model. Full details will be published separately.

The figure shows the co-ordination geometry with the iminoacyl-group bound as a *dihapto*-ligand. The Ru-C(2) and Ru-N distances are 2.031(8) and 2.123(7)  $\pounds$  respectively; the C(2)-N bond length of 1.30(1)  $\pounds$  is indistinguishable from an unco-ordinated CN double bond. The angles C(2)-N-C(4) and N-C(2)-C(3) have been





increased from those expected for a *monohapto*-iminoacyl ligand to 139.6(5) and 137.6(5) respectively as was also found in the similar complex  $Mo[n^2-C(NC_6H_5)CH_3](n^5-C_5H_5)(CO)_2$  (ref. 2).

The nitrogen atom of the *dihapto*-iminoacyl ligand is not readily alkylated, even with methyl triflate, but is rapidly protonated and reaction with HCl gives the amino-carbene complex, RuCl<sub>2</sub>[C(NHR)R](CO)(PPh<sub>3</sub>)<sub>2</sub>,VI.

## REFERENCES

1 W.R. Roper and L.J. Wright, J. Organometal. Chem., 142(1977)Cl. 2 R.D. Adams and D.F. Chodosh, Inorg. Chem., 17(1978)41.